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## **Atomic Structure Study Guide Answers**

Atomic Structures Study Guide

ANSWERS 1. The Periodic Table is mostly made of metals. 2. The noble gases are the farthest group to the right on the Periodic Table. 3. Atoms having equal numbers of positive and negative

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charges make them neutral. 4. The three primary sub-atomic particles that atoms are made of are protons, neutrons & electrons. 5.

## **Atomic Structures Study Guide ANSWERS**

Chemistry is the study of matter and the interactions between different types of

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matter and energy. The fundamental building block of matter is the atom. An atom consists of three main parts: protons, neutrons, and electrons. Protons have a positive electrical charge. Neutrons have no electrical charge. Electrons have a negative electrical charge.

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A charged atom. Dalton's Atomic Theory.

- 1) elements are made up of atoms
- 2) all atoms of a given element are identical
- 3) the atoms of an element are different from every other element
- 4) atoms of an element can combine to form a compound.

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laughed at. In the 1800's . John Dalton. proposed the idea of the "Atomic Theory". He had 5 theories, 3 of which are still believed today.

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Structures Study Guide ANSWERS 1. The Periodic Table is mostly made of metals.

2. The noble gases are the farthest group to the right on the Periodic Table.

3. Atoms having equal numbers of positive and negative charges make

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them neutral. 4.

## **Chapter 4 Atomic Structure Review Sheet Answer Key**

matter is made up of small particles called atoms with spaces in between them. atoms are neither created nor destroyed in chemical reaction merely rearranged. compounds always contain

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the same relative numbers and kinds of atoms. particles of atoms of an element are identical to each other.

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the number of each subatomic particle (electron, protons and neutrons) given the chemical. symbol of an atom. 4. charge, mass and location of the electron, proton and neutron.

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Unit 3 Study Guide: Atomic Structure

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and Nuclear Energy Section 4.1 Early Ideas About Matter For each statement below, write true or false. \_\_\_\_\_ 1.

Ancient philosophers regularly performed controlled experiments. \_\_\_\_\_

2. Philosophers formulated explanations about the nature of matter based on their own experiences.



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Answer and Explanation: Atomic structure is the composition of an element where there are negatively charged electrons circling around a positively charged nucleus. Chemical bonding is the...

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Unit 2 Atomic Structure & Periodic Table  
Study Guide Answer Section MULTIPLE  
CHOICE 1. ANS: A PTS: 1 2. ANS: B PTS:  
1 3. ANS: D PTS: 1 4. ANS: C PTS: 1 5.  
ANS: C PTS: 1 6. ANS: B PTS: 1 7. ANS: C  
PTS: 1 8. ANS: D PTS: 1 SHORT ANSWER  
9. ANS: Atomic Structure is the number

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of protons found in the nucleus.

## **Mrs. McCutchen's Class - Home**

34 Study Guide for An Introduction to Chemistry Exercise 3.3 - Cations and Anions: Identify each of the following as a cation or an anion, and determine the charge on each. (Obj 22) a. magnesium atom with 12 protons and 10 electrons

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$(+12) + (10) = +2$  This is a +2 cation.  
b. fluorine atom with 9 protons and 10 electrons

## **Chapter 3 The Structure of Matter and the Chemical Elements**

An atom is the fundamental building block of all matter. Everything in the universe is composed of individual

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atoms of various elements that combine together to form molecules and more complex structures. The basic structure of an atom consists of a nucleus containing protons and neutrons and a cloud of electrons that circle the nucleus. Part 1

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atom. The number after each elemental name represents the mass of the atom.

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## Homepage

www.njctl.org Chemistry Atomic Structure Answers 1 In the nuclear model of the atom, protons (and neutrons) are housed in a small, dense nucleus. Electrons surround the nucleus in an area of mostly empty space. 2 If electrons are electrically attracted to nucleus and would, therefore, have

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